

ADVANCED ENERGY MATERIALS

Supporting Information

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Unlocking Electrode Performance of Disordered Rocksalt Oxides Through Structural Defect Engineering and Surface Stabilization with Concentrated Electrolyte

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Supporting Figures

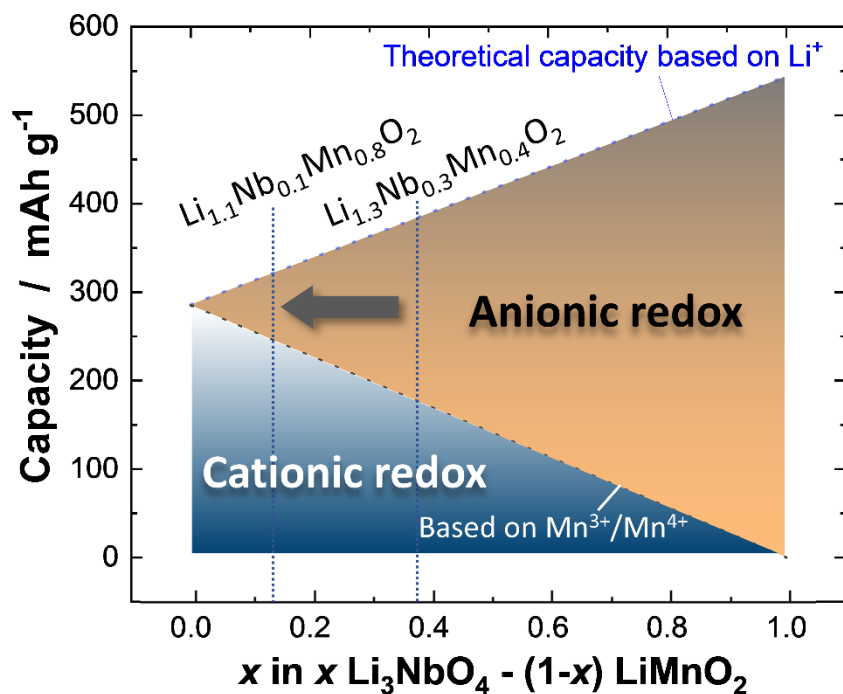


Figure S1. Theoretical capacities of Li₃NbO₄–LiMnO₂ binary systems with different chemical compositions. Theoretical capacities estimated based on the cationic (Mn³⁺/Mn⁴⁺ redox) and anionic (O²⁻/Oⁿ⁻ redox) are also shown.

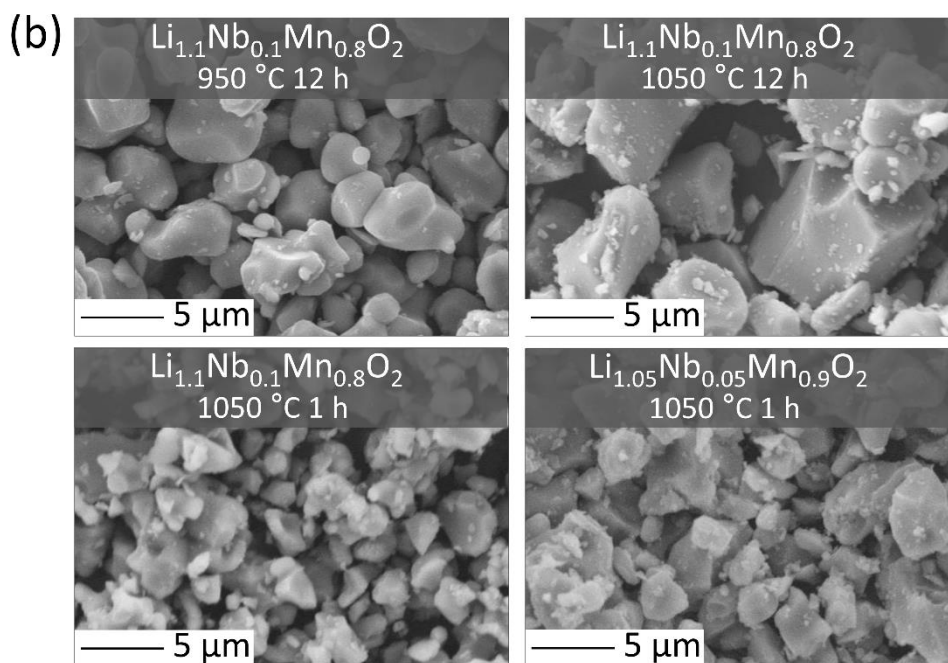
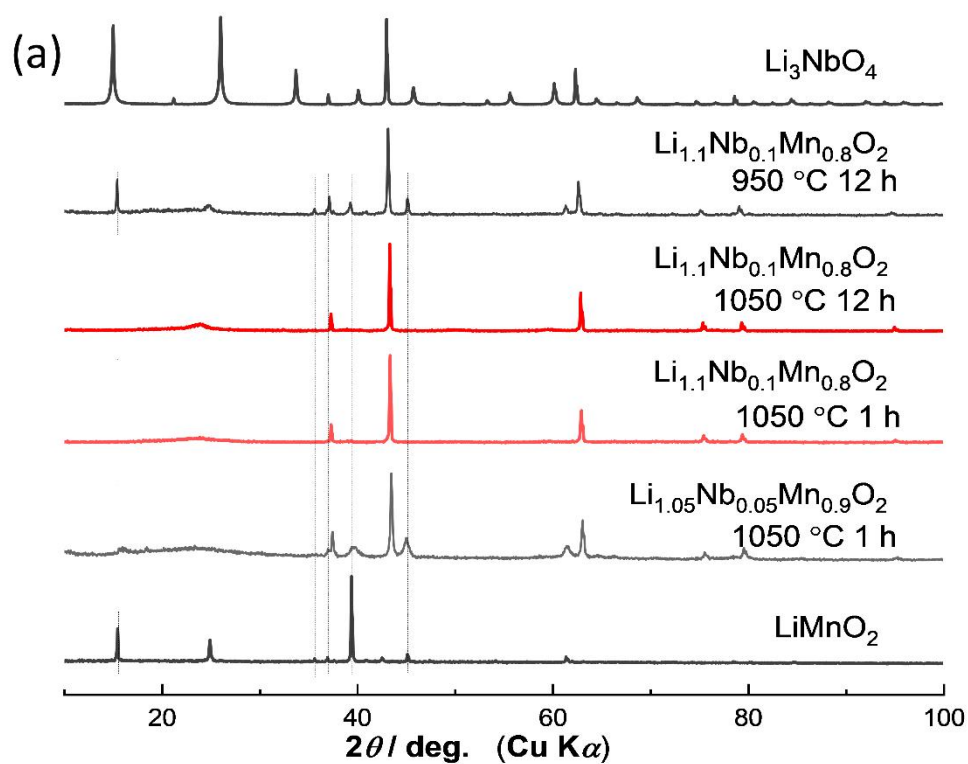


Figure S2. (a) XRD patterns and (b) SEM images of $\text{Li}_{1.1}\text{Nb}_{0.1}\text{Mn}_{0.8}\text{O}_2$ and $\text{Li}_{1.05}\text{Nb}_{0.05}\text{Mn}_{0.9}\text{O}_2$ synthesized at different temperature and time. The data of Li_3NbO_4 and LiMnO_2 are also shown for comparison. A pure phase of $\text{Li}_{1.05}\text{Nb}_{0.05}\text{Mn}_{0.9}\text{O}_2$ cannot be obtained.

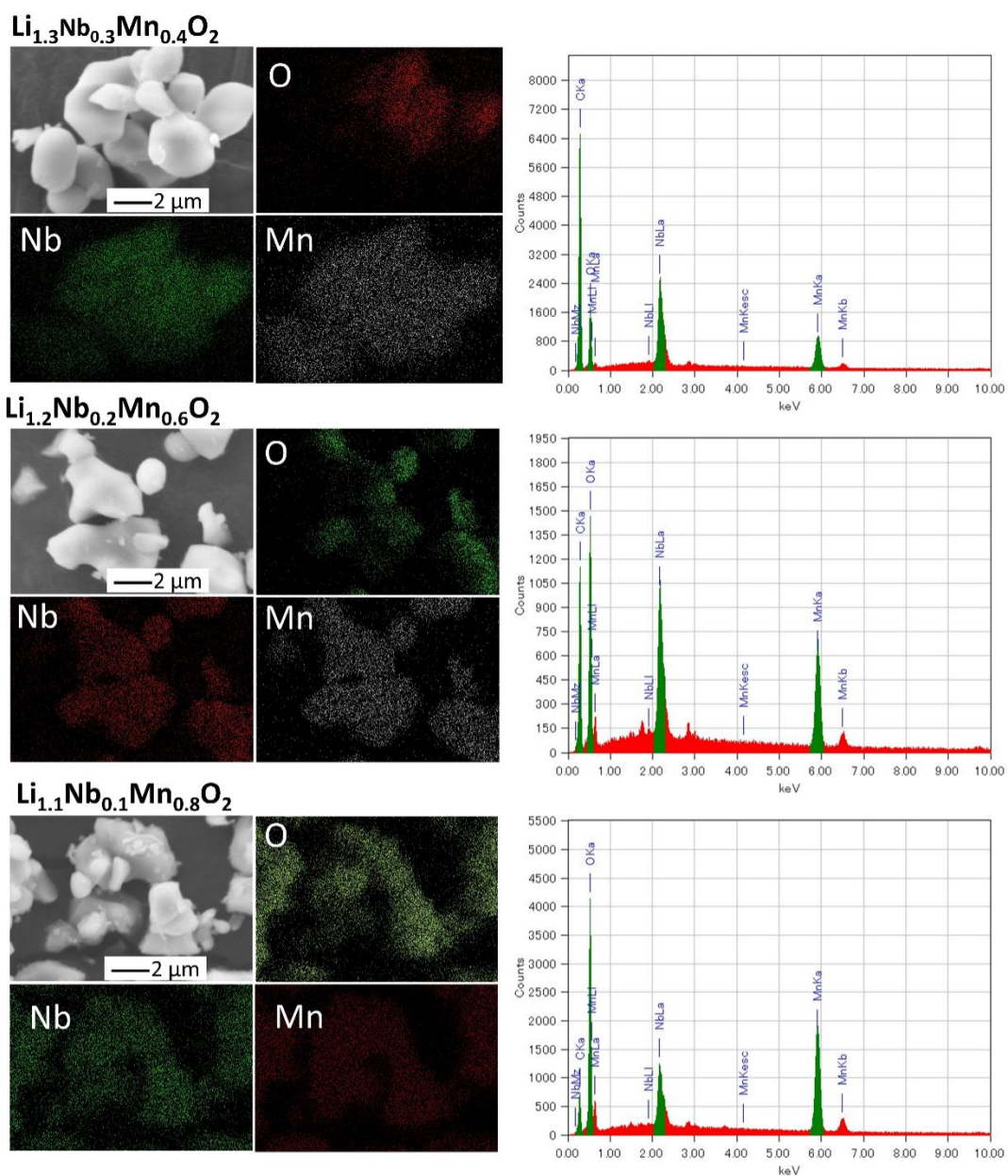


Figure S3. SEM images with EDX analysis of Li_{1.3}Nb_{0.3}Mn_{0.4}O₂, Li_{1.2}Nb_{0.2}Mn_{0.6}O₂, and Li_{1.1}Nb_{0.1}Mn_{0.8}O₂.

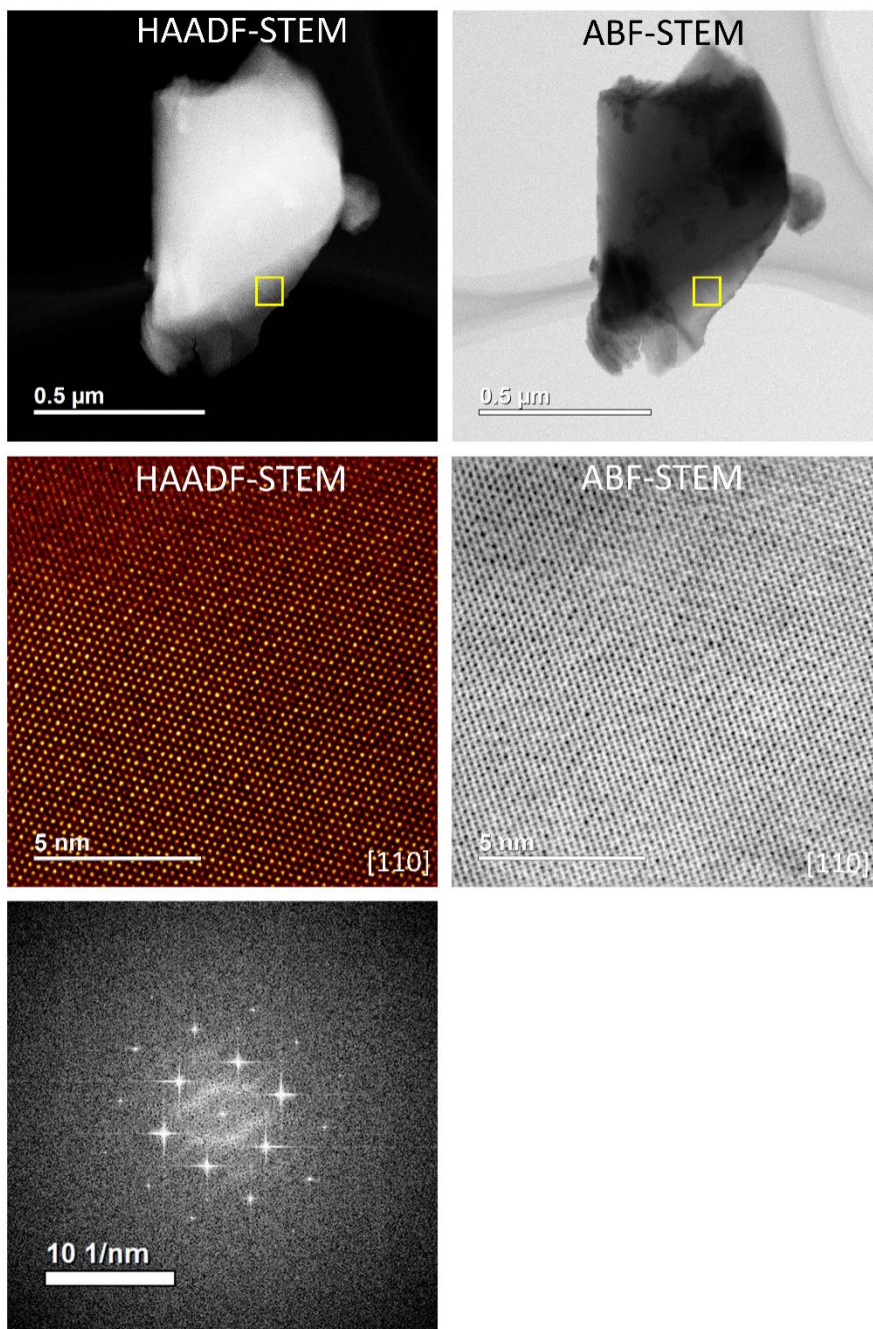


Figure S4. HAADF-/ABF-STEM images of $\text{Li}_{1.1}\text{Nb}_{0.1}\text{Mn}_{0.8}\text{O}_2$ along $[011]$, and high magnification images are shown in **Figure 1c**. An FFT image is obtained in the yellow square area, and some diffuse spots, which are indicative of short-range cation ordering, are observed (also see **Figure 1e** and **Supporting Figure S5**).

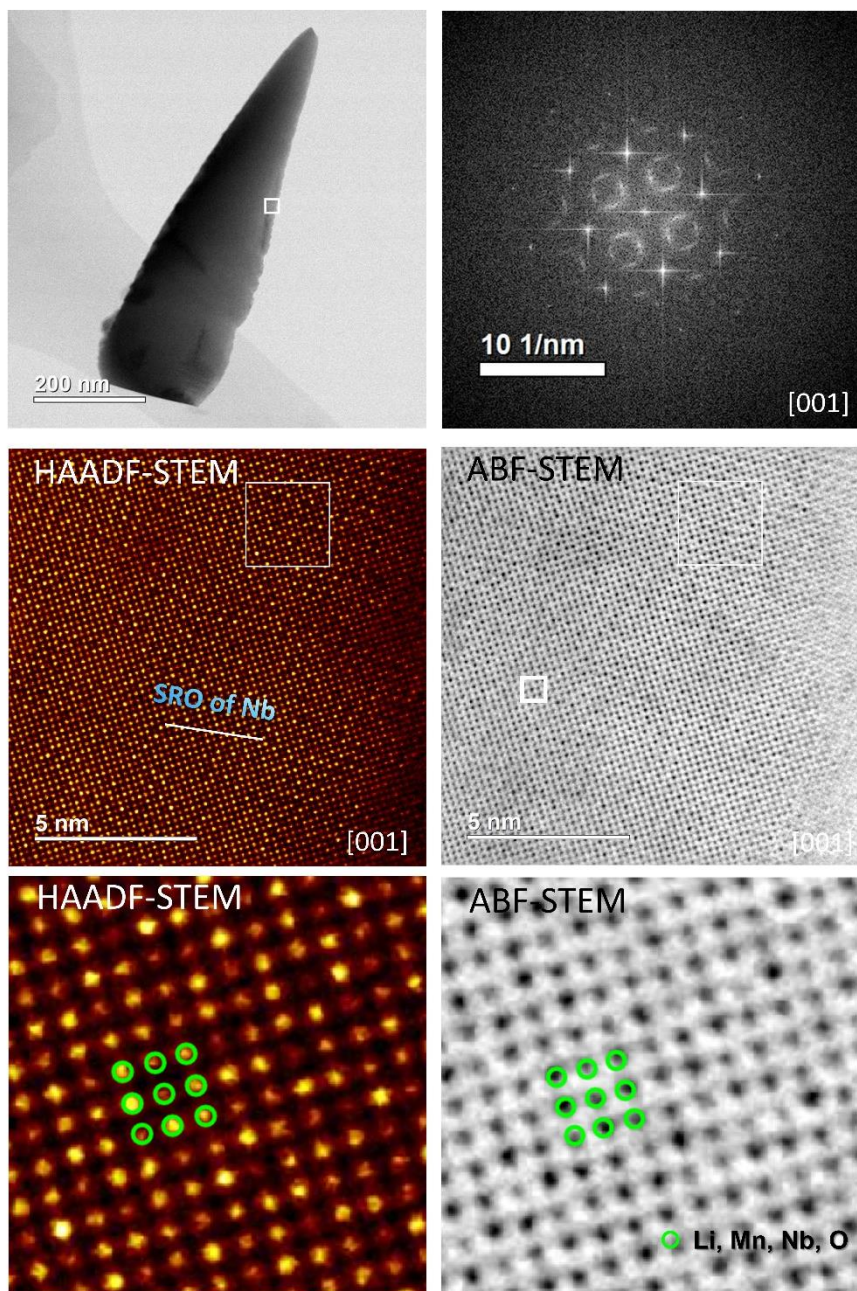


Figure S5. HAADF/ABF-STEM images of $\text{Li}_{1.1}\text{Nb}_{0.1}\text{Mn}_{0.8}\text{O}_2$ along [001] and an FFT image in the white square area in the low magnification image. Li, Nb, Mn, and O ions are randomly scattered along [001] zone axis, and the clear evidence of SRO is observed in the STEM image along [001].

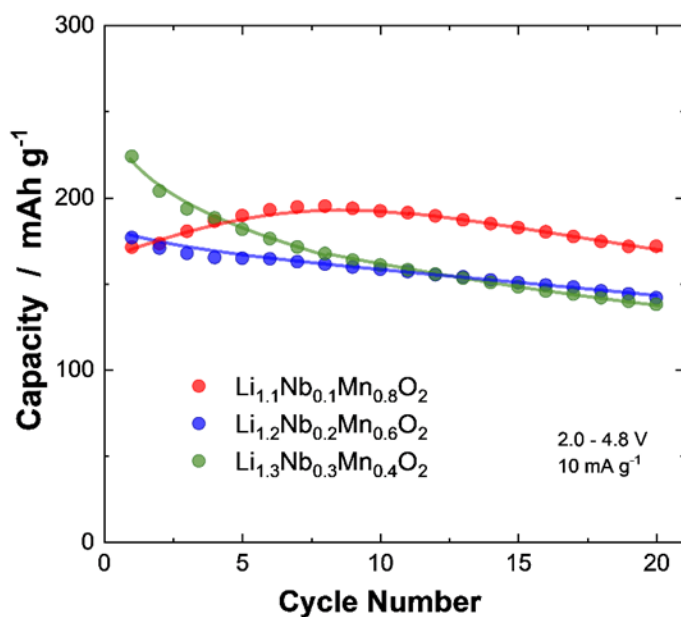


Figure S6. Cycle performance of $\text{Li}_{1.3}\text{Nb}_{0.3}\text{Mn}_{0.4}\text{O}_2$, $\text{Li}_{1.2}\text{Nb}_{0.2}\text{Mn}_{0.6}\text{O}_2$, and $\text{Li}_{1.1}\text{Nb}_{0.1}\text{Mn}_{0.8}\text{O}_2$ with a voltage range of 2.0 – 4.8 V at 10 mA g^{-1} .

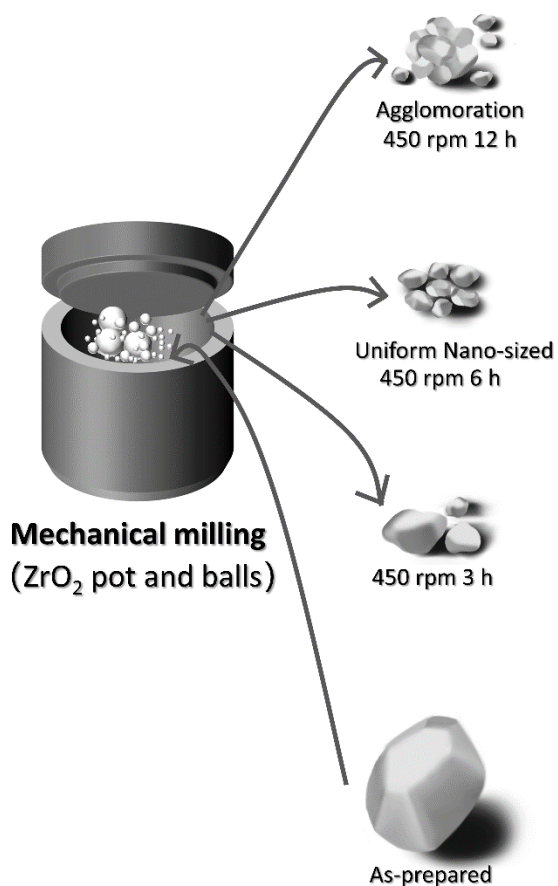


Figure S7. A schematic illustration of the synthesis of nanosized samples by mechanical milling.

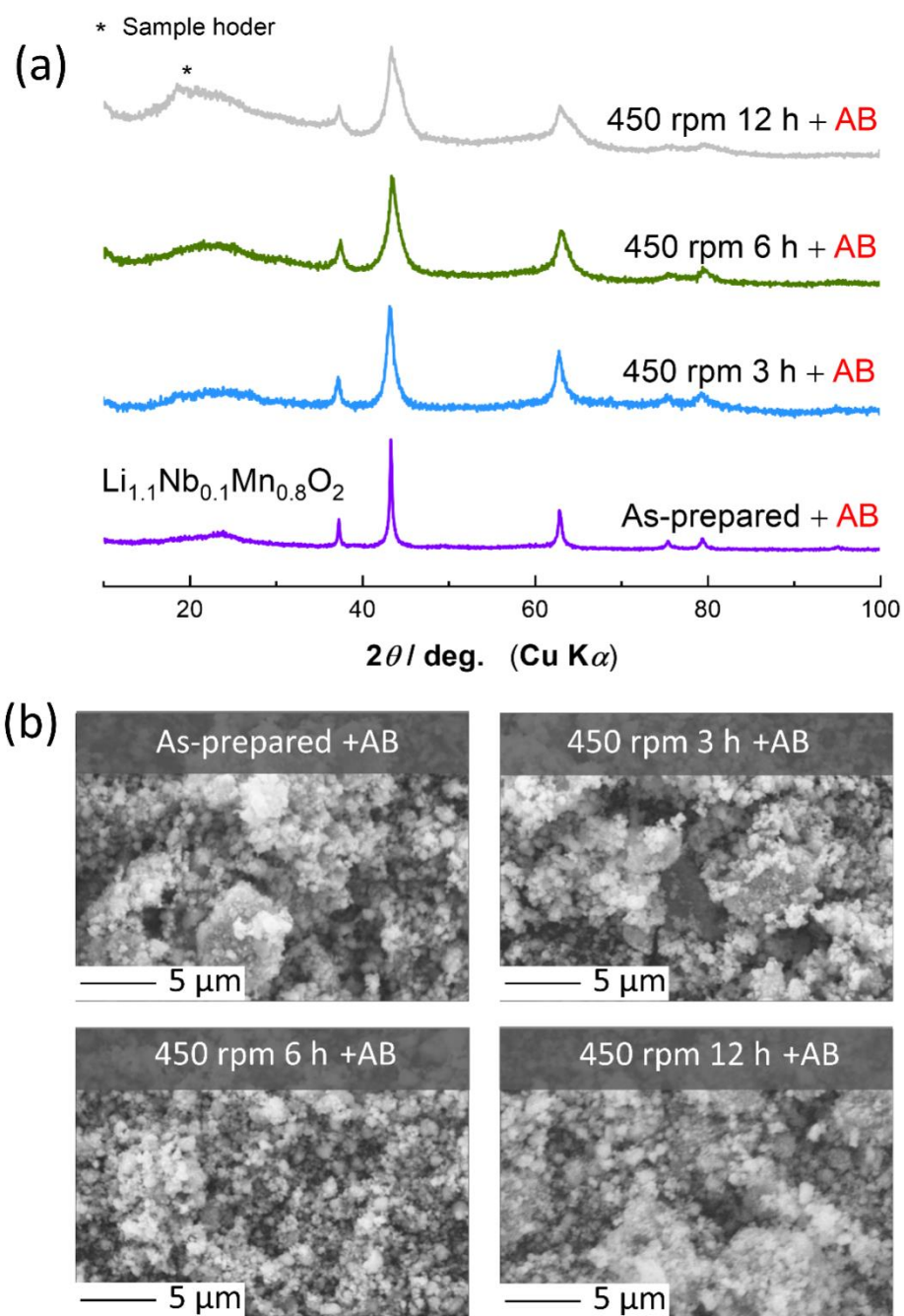


Figure S8. (a) XRD patterns and (b) SEM images of as-prepared $\text{Li}_{1.1}\text{Nb}_{0.1}\text{Mn}_{0.8}\text{O}_2$ samples milled at 450 rpm for 3 h, 6 h, and 12 h, after mixing with acetylene black.

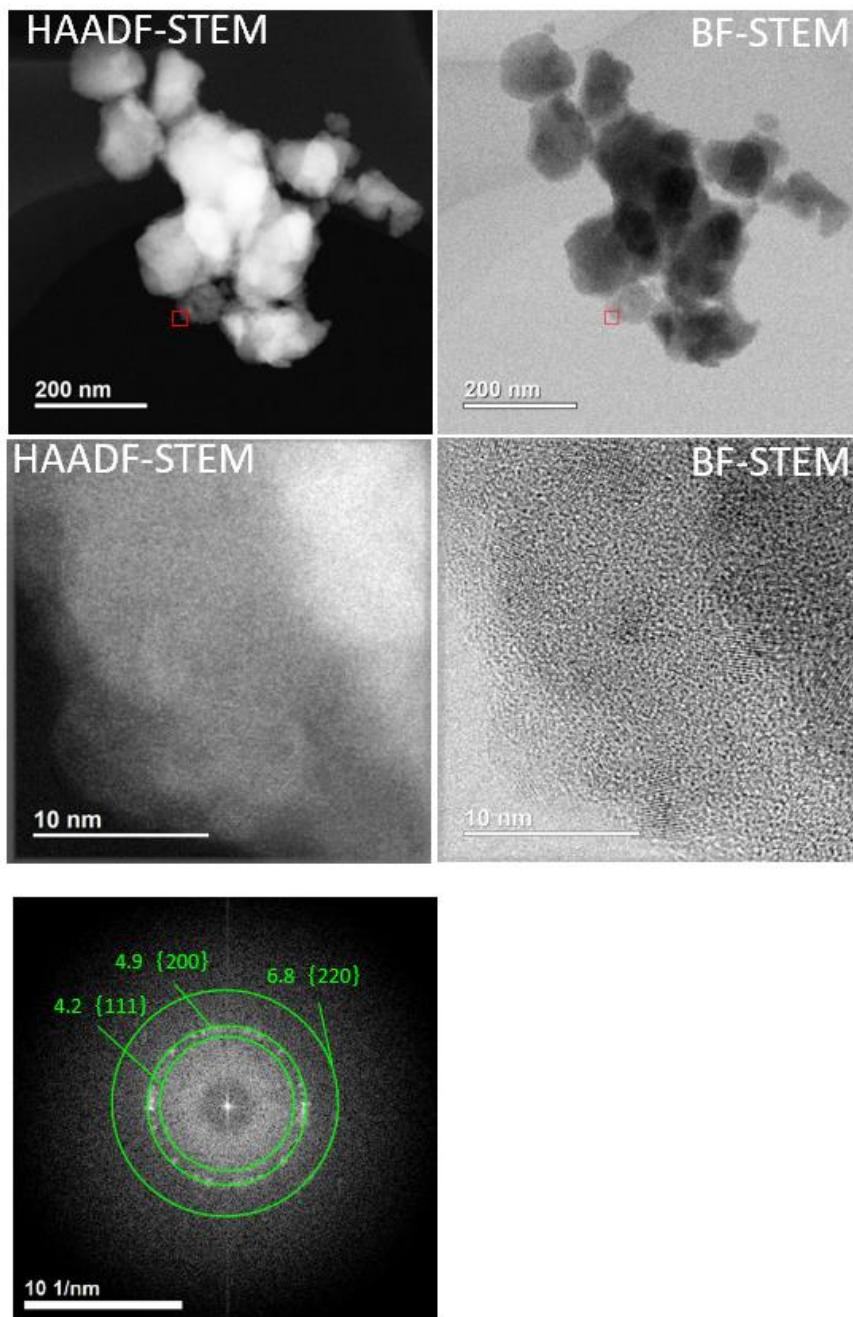


Figure S9. HAADF/ABF-STEM images of nanosized $\text{Li}_{1.1}\text{Nb}_{0.1}\text{Mn}_{0.8}\text{O}_2$, 450 rpm 6 h milled sample, with different magnifications. An FFT image of STEM image is also shown.

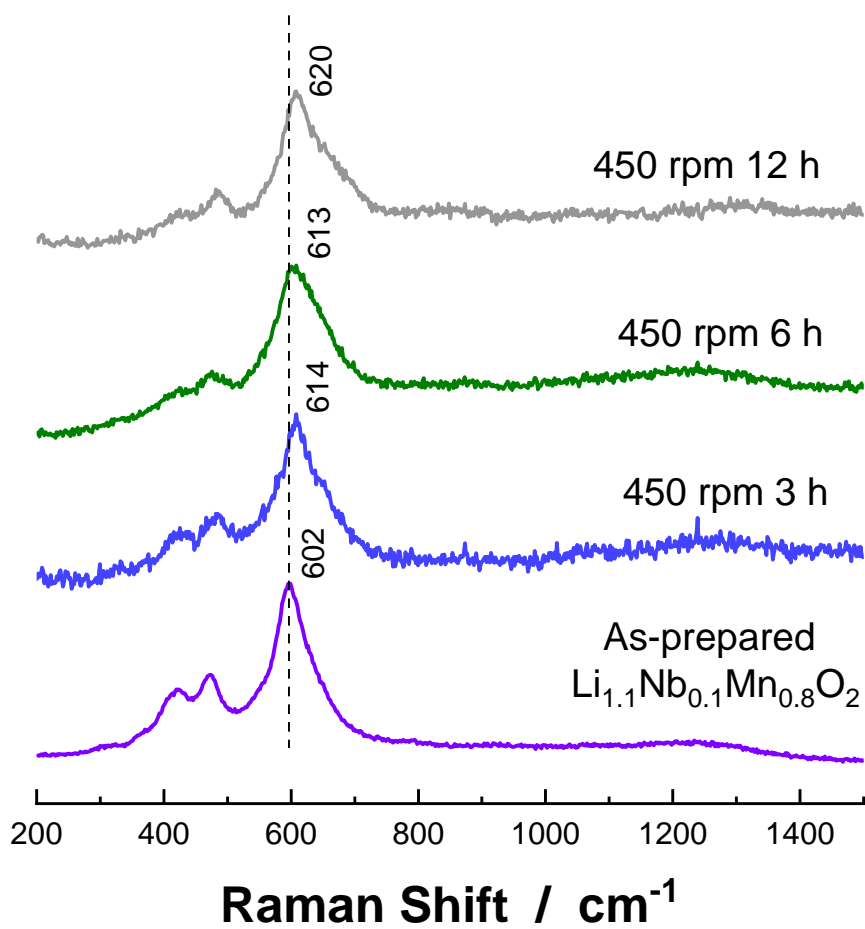


Figure S10. Raman spectra of as-prepared, 450 rpm 3 h, 6 h, and 12 h milled samples.

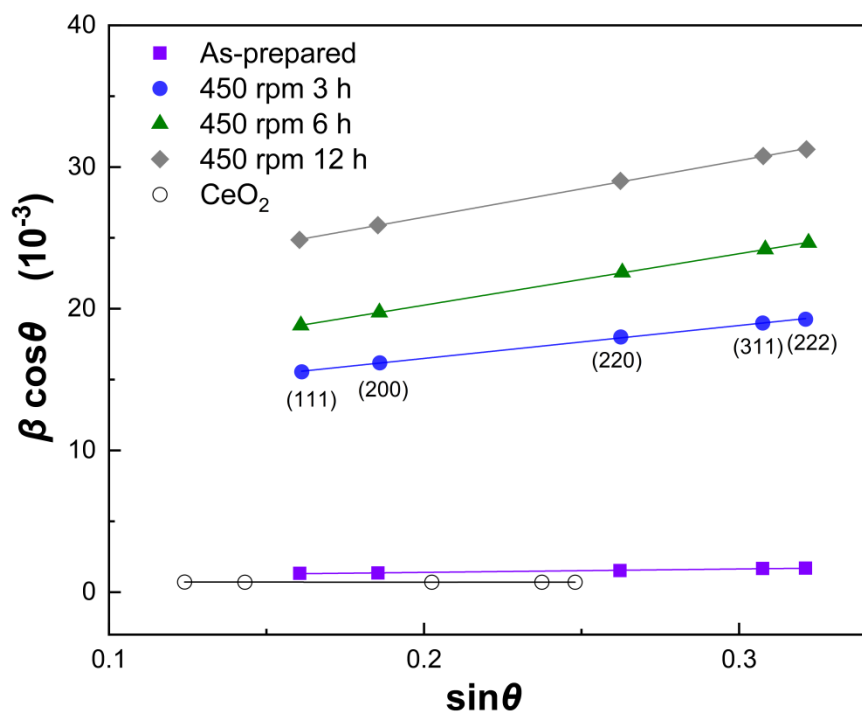


Figure S11. Williamson–Hall plots of the different samples: as-prepared, 450 rpm 3h, 6 h, and 12 h. A CeO_2 standard was also analyzed for instrumental calibration.

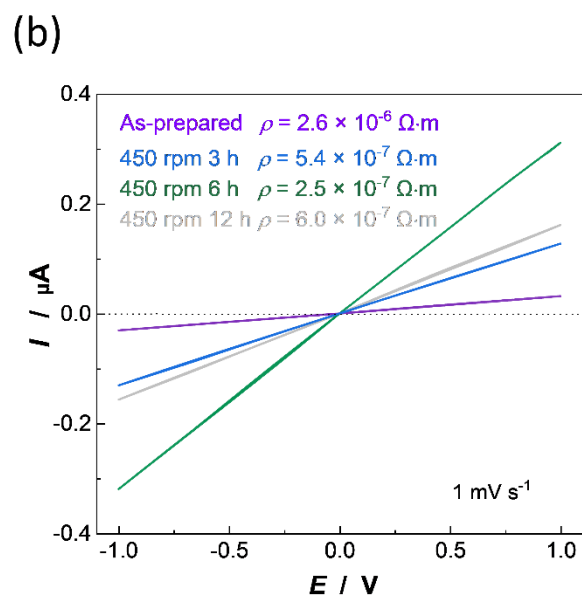
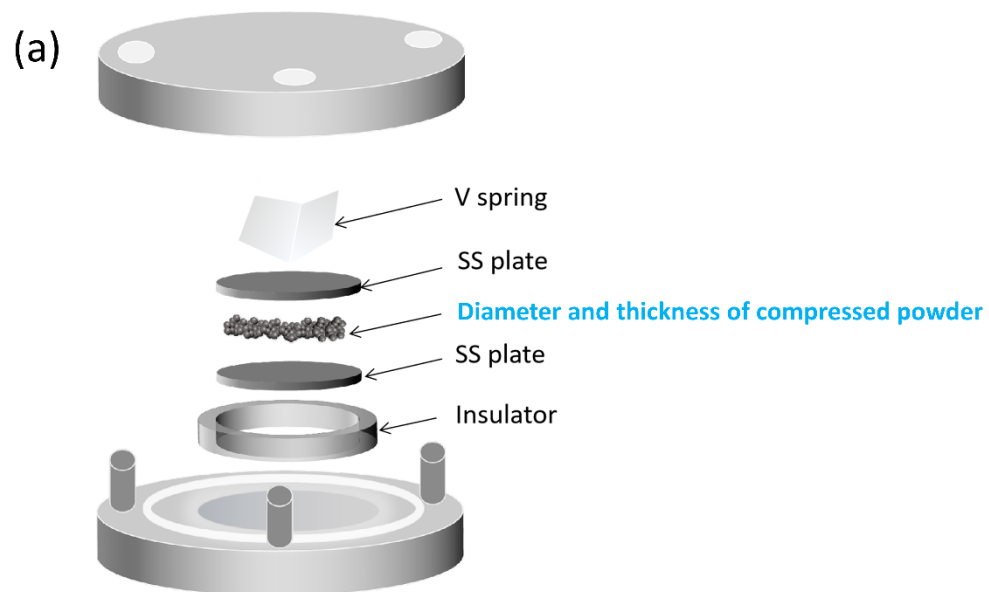


Figure S12. (a) A scheme of the experimental setup of electronic conductivity measurement for powder samples, and (b) reproduced data obtained from different powders synthesized at the same condition in **Figure 3c**.

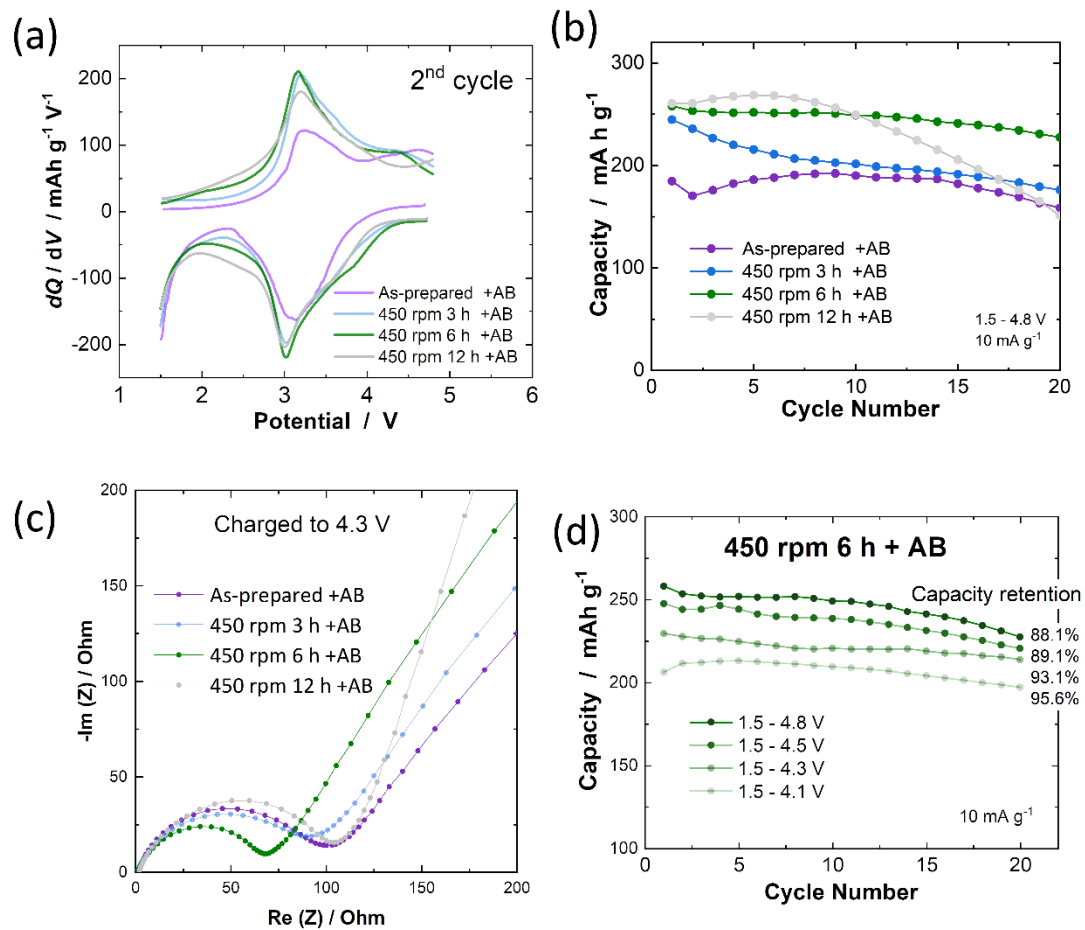


Figure S13. Comparison of electrode performance before and after mechanical milling with different duration: (a) differential capacity curves, (b) cycle stability, (c) EIS spectra after charged to 4.3 V, and (d) cycle performance of 450 rpm 6 h milled $\text{Li}_{1.1}\text{Nb}_{0.1}\text{Mn}_{0.8}\text{O}_2$ at different cut-off voltages.

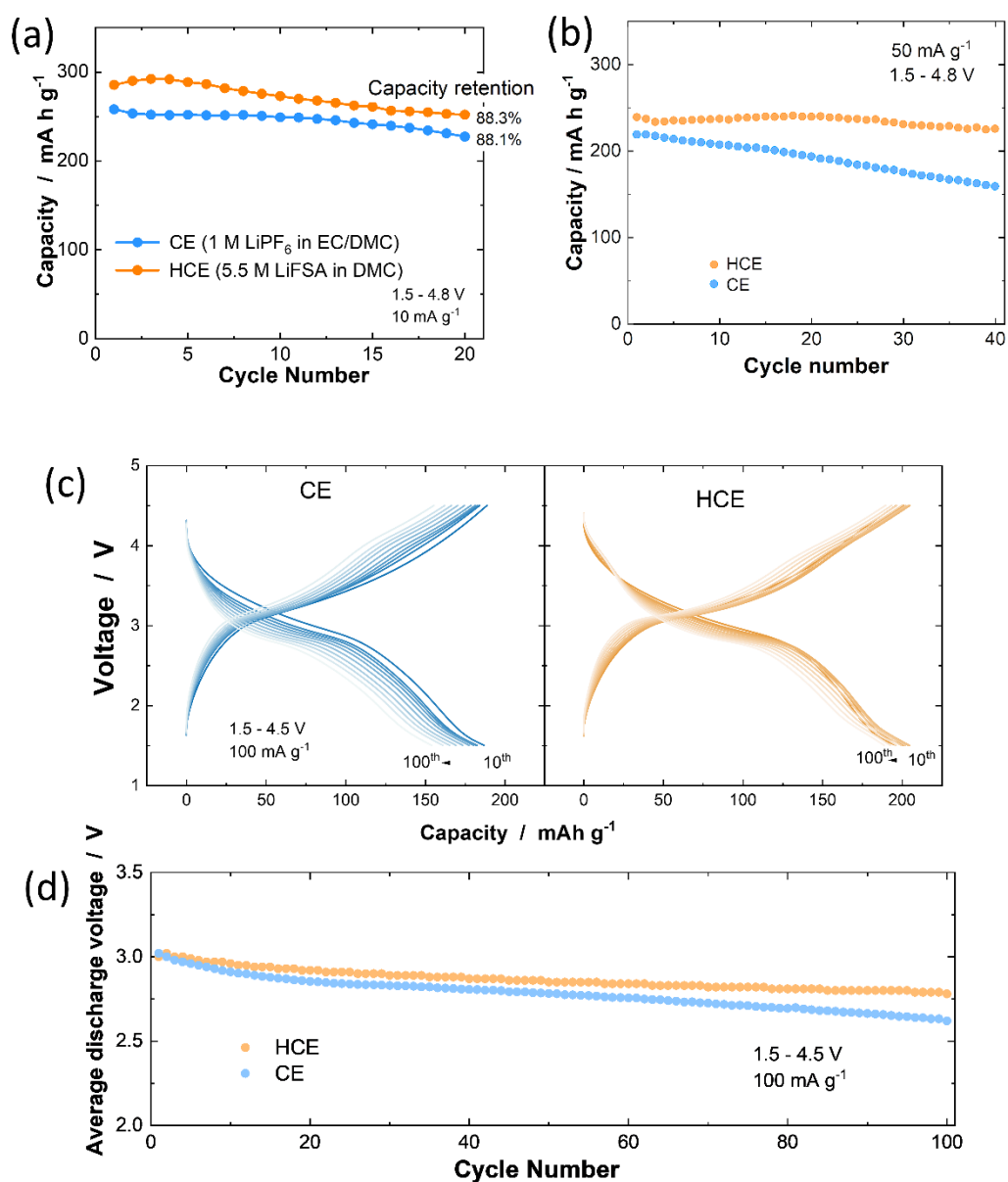


Figure S14. Electrode performance of nanosized $\text{Li}_{1.1}\text{Nb}_{0.1}\text{Mn}_{0.8}\text{O}_2$, 450 rpm 6 h milled sample, cycled in CE and HCE: cyclability at a rate of (a) 10 mA g^{-1} and (b) 50 mA g^{-1} with the voltage range of $1.5-4.8 \text{ V}$. (c) Galvanostatic charge/discharge curves cycling in CE and HCE at 100 mA g^{-1} with voltage range of $1.5-4.5 \text{ V}$ and (d) average discharge voltage variations for 100 cycles.

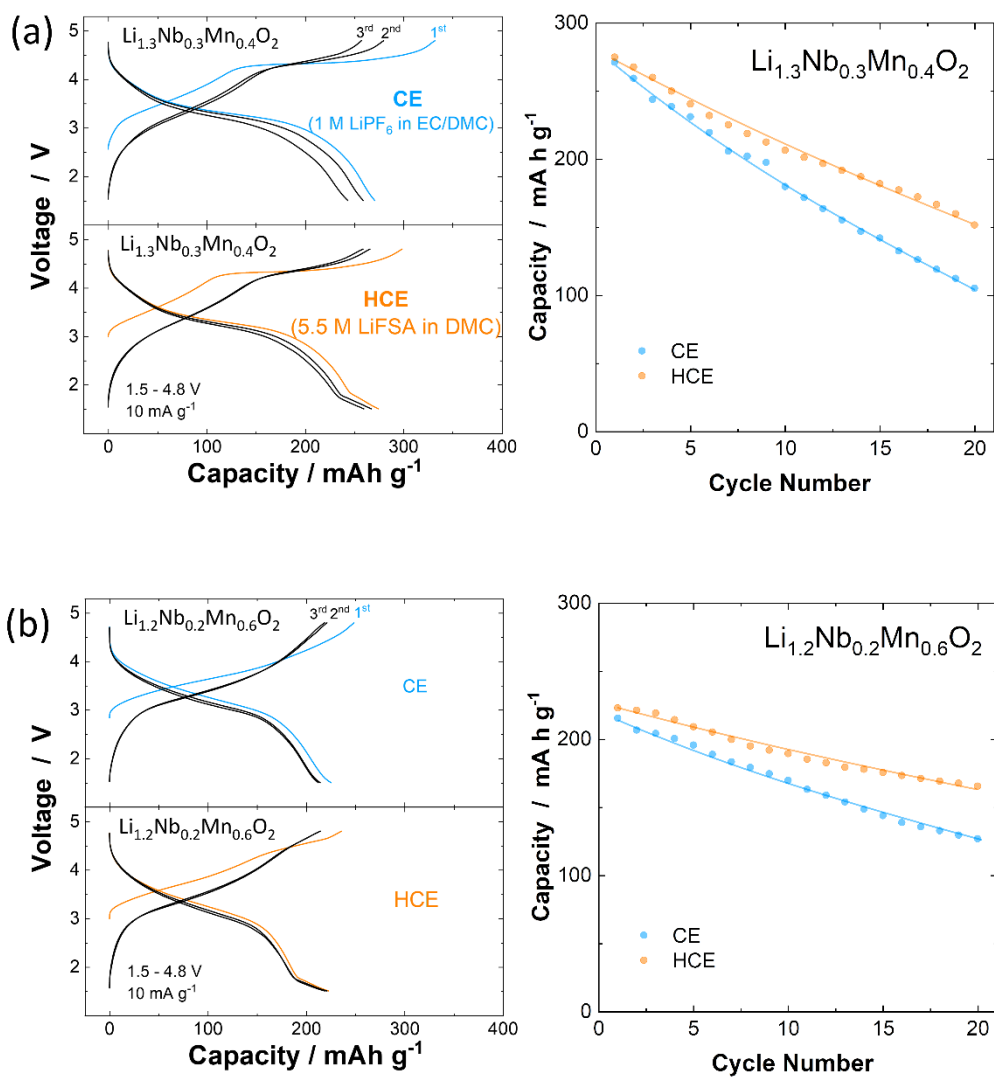


Figure S15. Electrode performance comparison of micrometer-sized (a) $\text{Li}_{1.3}\text{Nb}_{0.3}\text{Mn}_{0.4}\text{O}_2$ and (b) $\text{Li}_{1.2}\text{Nb}_{0.2}\text{Mn}_{0.6}\text{O}_2$ cycled in CE and HCE.

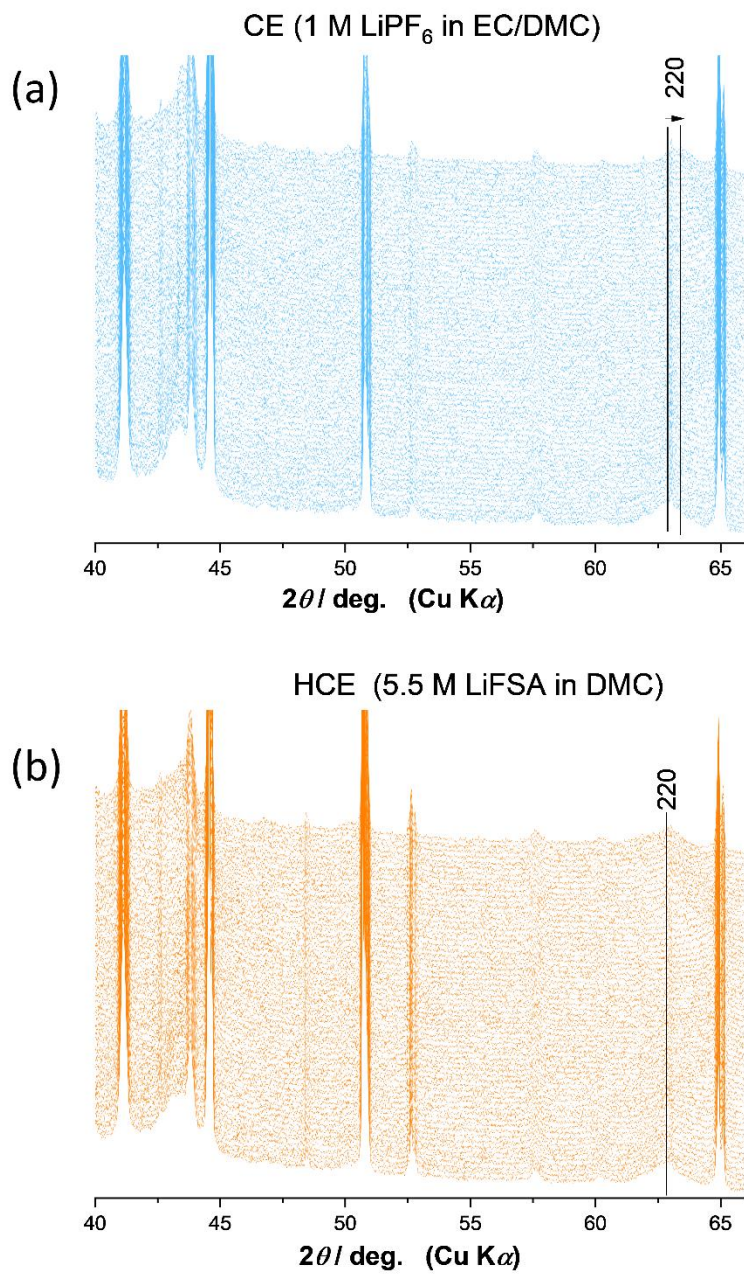


Figure S16. Original in-situ XRD data of nanosized Li_{1.1}Nb_{0.1}Mn_{0.8}O₂ cycled in (a) CE and (b) HCE. Many diffraction peaks originate from Be window and Al current collector.

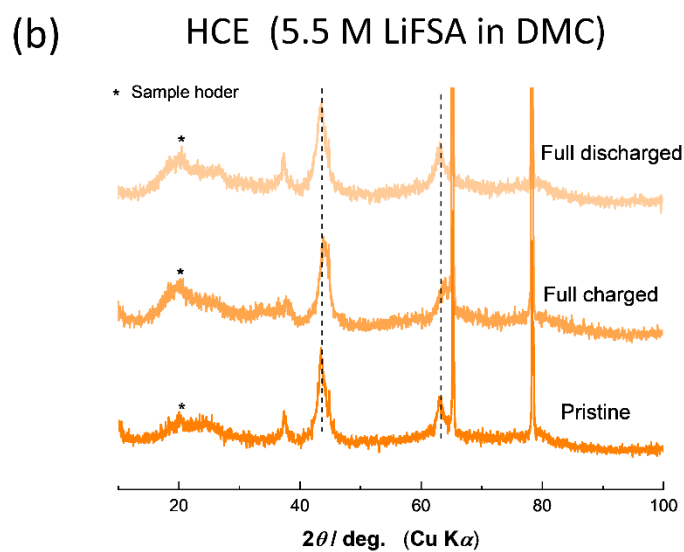
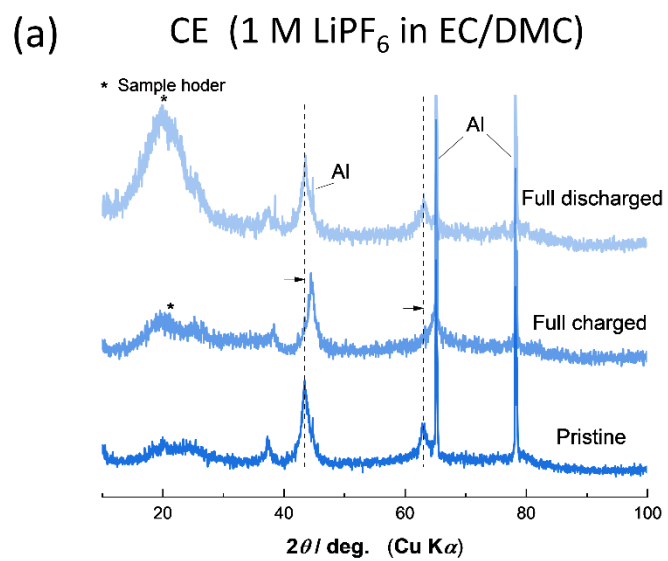


Figure S17. Ex-situ XRD data of nanosized Li_{1.1}Nb_{0.1}Mn_{0.8}O₂ cycled in (a) CE and (b) HCE.

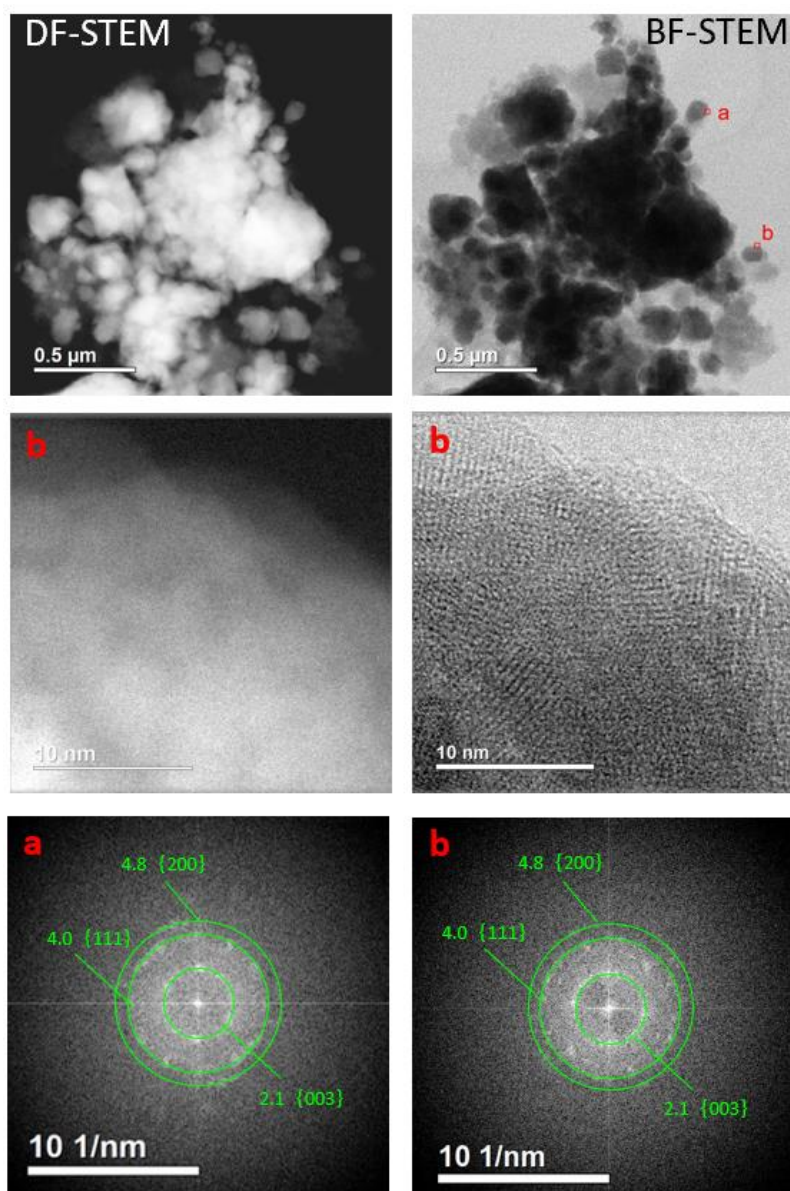


Figure S18. DF/BF-STEM and FFT images of nanosized $\text{Li}_{1.1}\text{Nb}_{0.1}\text{Mn}_{0.8}\text{O}_2$ cycled in HCE with different magnifications. The red square “a” is the location for STEM images shown in **Figure 6b**.

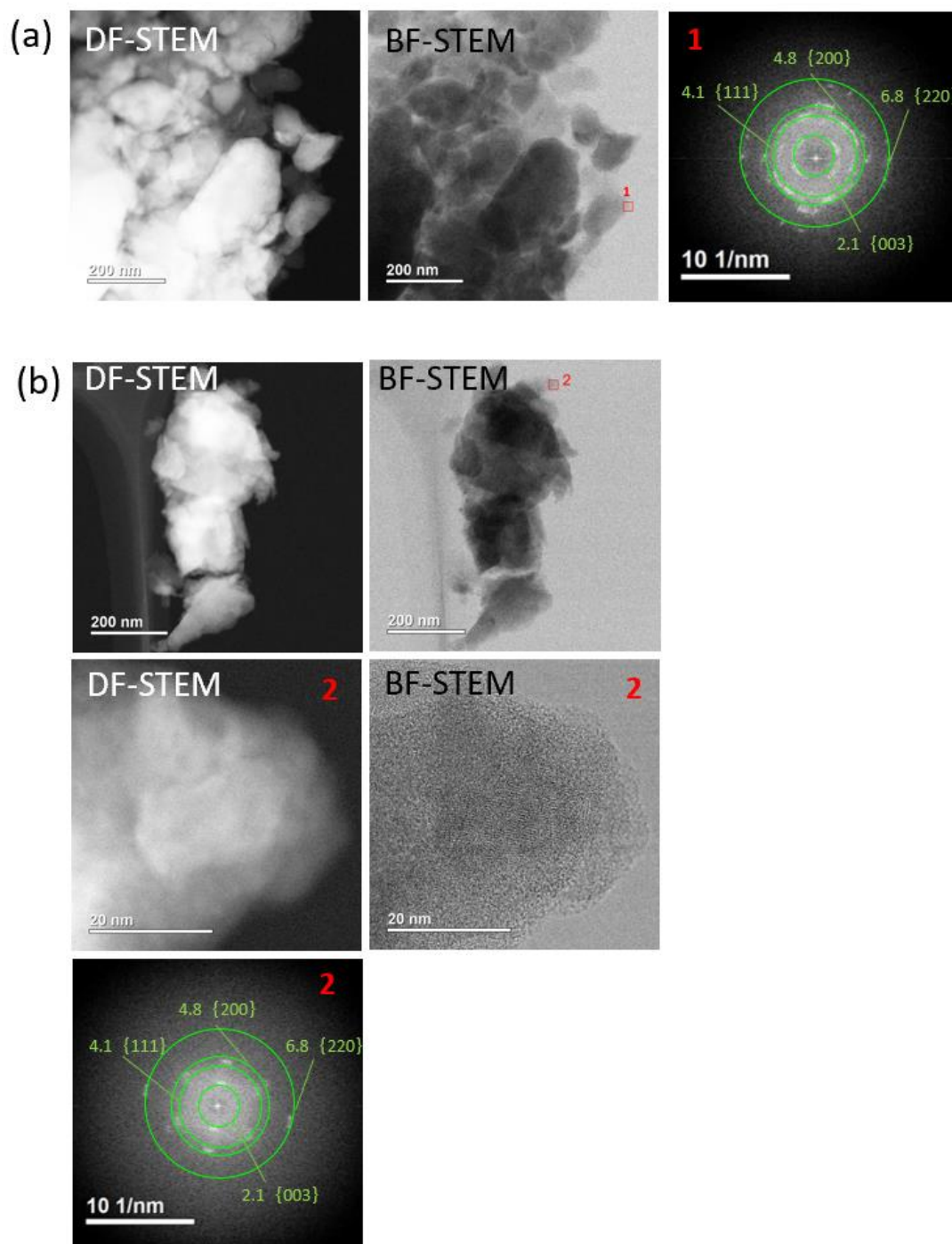


Figure S19. DF/BF-STEM and FFT images of (a) nanosized $\text{Li}_{1.1}\text{Nb}_{0.1}\text{Mn}_{0.8}\text{O}_2$ cycled in CE, the red square “1” is the location for STEM images shown in **Figure 6b**, and (b) the sample measured from a different particle.

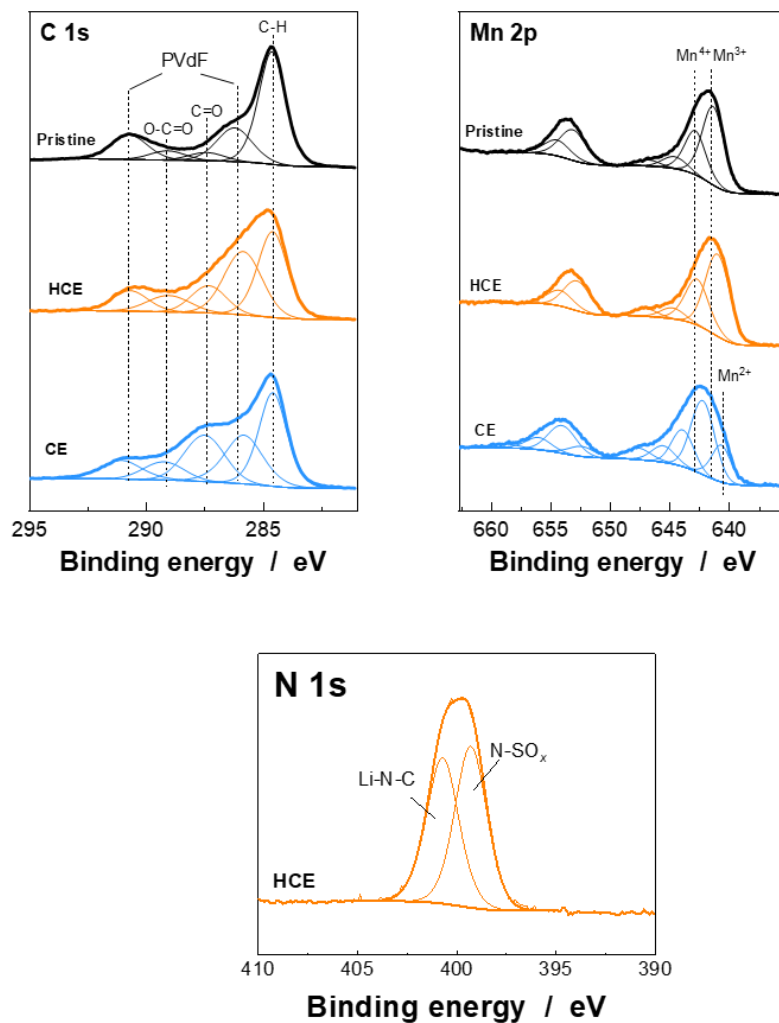


Figure S20. XPS spectra of nanosized $\text{Li}_{1.1}\text{Nb}_{0.1}\text{Mn}_{0.8}\text{O}_2$ before and after cycle in HCE and CE: C 1s, Mn 2p, and N 1s XPS spectra.